# PREPARATION OF <sup>18</sup>0-BILABELLED CARBOXYLIC ACIDS: CINNAMIC AND PHENYLPROPIOLIC ACIDS via STYRYLISOXAZOLES

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#### SUMMARY

3-Pheny1-2-propenoic (cinnamic) and 3- $[^2H_5]$ -pheny1-2-propenoic  $^{18}$ 0-bilabelled acids ( $\underline{2}$ c,d) are prepared  $\underline{via}$  alkaline hydrolysis of styrylisoxazoles ( $\underline{1}$ a,b). The same reaction carried out on the  $\underline{vic}$ -dibromocompounds ( $\underline{3}$ a,b) gave the 3-pheny1-2-propynoic (phenylpropiolic) and 3- $[^2H_5]$ -pheny1-2-propynoic  $^{18}$ 0-bilabelled acids (4c,d), respectively.

### KEY WORDS

Oxygen-18 and deuterium labelled compounds; 3-phenyl-2-propenoic and 3-phenyl-2-propynoic acids.

#### INTRODUCTION

The importance of isoxazole derivatives as intermediates in organic synthesis is well established on the basis of the current literature on this topic (1-4). Moreover syntheses of natural

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products or related compounds, e.g. steroids (5) and  $(\pm)$ -[6]-gingerol (6) and corrins (7) were achieved by following strategies which involved isoxazole derivatives as a part of the synthetic routes.

Under the former line we have already reported the preparation of cinnamic (8), coumaric (9) and phenylpropiolic (10) acids  $\underline{via}$  styrylisoxazoles. The use of Na<sup>18</sup>OH (<sup>18</sup>O 20%) for the cleavage of the isoxazole ring in the preparation of cinnamic acid, allowed us to clarify the role of nucleophilic attack of the alkaline medium on the  $C_5$  of the isoxazole ring which in turn became the carbon atom of the carboxylic group (11). In addition, <sup>18</sup>Obilabelled coumaric acid was also obtained by this route (9). Because the methods reported in the literature for the preparation of the <sup>18</sup>O-bilabelled carboxylic acids mainly involve the use of  $C^{18}O_2(12)$ , hydrolysis, or exchanging procedures with  $H_2^{18}O$  (13), we found it interesting to extend our previous experiments to the preparation of cinnamic and phenylpropiolic acids with two

### RESULTS AND DISCUSSION

The styrylisoxazoles ( $\underline{1}a$ ,b) used in this study as starting materials were obtained according to the literature (14) by condensing 3,5-dimethyl-4-nitroisoxazole with benzaldehyde and  $[^2H_5]$ -benzaldehyde respectively. Alkaline hydrolysis (NaOH 1N) of compounds  $\underline{1}a$  and  $\underline{1}b$ , followed by acidification, gave the expected cinnamic acids  $\underline{2}a$  (8) and  $\underline{2}b$  which showed peaks in the mass spectra at m/e=148, 147 and m/e=153, 152, respectively. These values were attributed to the corresponding molecular ions M<sup>+</sup> and M<sup>+</sup>+1. When the same hydrolysis was carried out with Na $^{18}OH$  ( $^{18}OH$ )

99.8%), the mass spectra of the resulting acids showed signals at m/e=152, 151 and m/e=157,156. These findings were in agreement for M  $^+$  and M $^+$ -1 of the expected  $^{18}$ O-bilabelled cinnamic and 3-  $^{2}$ H<sub>5</sub>-phenyl-2-propenoic acids  $^{2}$ C and  $^{2}$ d.

Alkaline hydrolysis (NaOH 1N) followed by acidification of  $\underline{3}$ a (10) and  $\underline{3}$ b gave the 3-phenyl- and  $3-[^2H_5]$ - phenyl-2-propynoic acids  $\underline{4}$ a and  $\underline{4}$ b, which showed signals in the mass spectra at m/e=146 (M<sup>+</sup>) and m/e=151 (M<sup>+</sup>).

H<sub>3</sub>C NO<sub>2</sub>
H<sub>2</sub>CH-CH-Ar
Br Br

1
3
$$|i00H^{-or}|^{18}OH^{-}$$

$$|iii|H^{+}$$

$$Z$$
C=CAr
$$2$$

$$\frac{1}{2}a, \frac{3}{2}a$$

$$\frac{1}{2}b, \frac{3}{2}b$$

$$\frac{1}{2}b, \frac{4}{2}b$$

$$\frac{1}{2}c, \frac{4}{2}c$$

$$\frac{$$

Treatment of  $\underline{3}a$  and  $\underline{3}b$  with  $\underline{N}a^{18}OH$  ( $^{18}O$  99%) afforded  $\underline{4}c$  and  $\underline{4}d$ , whose peaks in the mass spectra appeared at m/e=150 (M<sup>+</sup>) and m/e=155 (M<sup>+</sup>). These values are consistent with those expected for

the molecular ion of  $^{18}$ O-bilabelled 3-phenyl- and  $^{2}$ H<sub>5</sub>]-phenyl-2-propynoic acids  $^{4}$ c and  $^{4}$ d, respectively. It is noteworthy to consider that the above method, which implies only the use of  $^{18}$ O as labelling reagent, is based on the alkaline hydrolysis of the 3-methyl-4-nitroisoxazol-5-yl group, which is a masked carboxylic group.

This reaction appears to be candidate as a general method for preparing  $^{18}\text{O-bilabelled}$  carboxylic acids.

### EXPERIMENTAL

Mass spectral data of compounds  $\underline{2}$ a-d and  $\underline{4}$ a-b were taken by direct inlet in the LKB mass spectrometer, whereas a VG 70-70 EQ spectrometer was used for those of compounds  $\underline{4}$ c-d.  $^{1}$ H-N.M.R. spectra were recorded on a Perkin-Elmer R32 spectrometer and reported in p.p.m. with tetramethylsilane as internal standard. Elemental analyses were performed with a Perkin-Elmer 240C apparatus; percentage of deuterium in compounds  $\underline{1}$ b and  $\underline{3}$ b, was calculated for the right M.W. and found as hydrogen. 3-Methyl-4-nitro-5-styrylisoxazole ( $\underline{1}$ a), cinnamic acid ( $\underline{2}$ a), 5-(1,2-dibromo-2-phenylethyl)-3-methyl-4-nitroisoxazole ( $\underline{3}$ a) and phenylpropiolic acid ( $\underline{4}$ a) were prepared according to the ref. (14), (8), (15) and (10), respectively. Na $^{18}$ OH was prepared by careful addition of Na to  $\mathrm{H}_2$  180 (isotopic enrichment was 99.8% and 99% for the hydrolysis of  $\underline{1}$ a-b and  $\underline{3}$ a-b, respectively).

3-Methyl-4-nitro-5-(2- $\begin{bmatrix} 2H_5 \end{bmatrix}$ -phenylethenyl)-isoxazole (1b)

Compound <u>1</u>b was prepared by condensing 3,5-dimethyl-4-nitroisoxazole with  $[^2H_5]$ -benzaldehyde (Merck,98%) following the procedure reported in the literature (14) for <u>1</u>a. Yield 80.6%, m.p. 154-155°C (from ethanol).

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N.M.R. (CDCl<sub>3</sub>): 2.57 (s, 3H, CH<sub>3</sub>); 7.69 (AB system, 2H,  $J_{AB}$ = 16.6Hz).

Anal. Calcd. for  $C_{12}D_5H_5N_2O_3$ : C, 61.26; H, 4.24; N, 11.95 Found: C, 61.39; H, 4.36; N, 12.15%.

## $3-[^2H_5]-Phenyl-2-propenoic acid (2b)$

Compound <u>1</u>b (g 0.350) treated as reported in ref. (9) for  $\underline{2}$ a gave  $\underline{2}$ b (g 0.148, 65% yield) which was purified by sublimation (110°C, 0.6 mmHg), m.p. 133°C.

### $18_{0-Bilabelled}$ 3-phenyl-2-propenoic acid (2c)

Compound  $\underline{1}a$  (g 0.030) treated with Na<sup>18</sup>OH 1N (ml 0.5) as reported in ref. (9) for  $\underline{2}a$ , gave  $\underline{2}c$ , which was purified by sublimation (100°C, 0.6 mmHg).

# $^{18}$ O-Bilabelled $^{3-}$ $^{2}$ H<sub>5</sub> $^{-}$ phenyl-2-propenoic acid (2d)

Compound 1b (g 0.013) treated with Na $^{18}$ OH (m1 0.5), was heated (95°C) under stirring for 4 h. The cold mixture was then filtered and the solution acidified with hydrochloric acid (gas) to give the expected 2d which was purified by sublimation (g 0.001; 100°C, 0.6 mmHg).

 $5-(1,2-Dibromo-2-[^2H_5]-phenylethyl)-3-methyl-4-nitroisoxazole$  (3b) Compound 3b was prepared by treating 1b (g 0.560) in CS<sub>2</sub> with bromine following ref.(11). Yield 71.8%, m.p. 168-169°C (from ethanol).

I.R.(KBr): 1615, 1520, 1415, 1379, 1365, 1300, 1140 and 822 cm<sup>-1</sup> N.M.R. (CDCl<sub>3</sub>): 2.63 (s, 3H, CH<sub>3</sub>); 5.55-6.45 (AB system, centered at 6.0,  $J_{AB}$  = 12Hz).

Anal. Calcd. for  $C_{12}D_5H_5N_2O_3Br_2$ : C, 36.47; H, 2.53; N, 7.09 Found: C, 36.76; H, 2.67; N, 7.11%.

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# $3-[^2H_5]-Phenyl-2-propynoic acid (4b)$

Compound 3b (g 0.400) was treated with NaOH 1N (ml 8) and heated (95°C) for 3 h. Unreacted 3b was filtered off (g 0.018) and the solution acidified with HCl (gas) to give the expected compound 4b (g 0.103) which was purified by sublimation (g 0.079; 55°C, 0.5 mmHg).

### 180-Bilabelled 3-phenyl-2-propynoic acid (4c)

Compound 3a (g 0.025) was treated with Na<sup>18</sup>0H 1N (ml 0.5) and heated (95°C) for 4 h. The cold mixture was then filtered and the solution acidified with hydrochloric acid (gas) to give the expected compound 4c which was purified by sublimation (g 0.001; 107°C; 0.5 mmHg).

# $^{18}$ 0-Bilabelled 3- $^{2}$ H<sub>5</sub>-phenyl-2-propynoic acid (4d)

Compound 3b (g 0.025) was treated with Na<sup>18</sup>0H 1N (ml 0.5) and heated (95°C) for 4 h. The cold mixture was then filtered and the solution acidified with hydrochloric acid (gas) to give the expected compound 4d which was purified by sublimation (g 0.001; 90°C; 0.5 mmHg).

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